



# Climate Action and the Nuclear Conundrum

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It may facilitate understanding of the three main arguments that underlie promotion of the nuclear option to review the basics of atomic structure and nuclear power plant operation, so I begin with a little background. Readers wishing to by-pass this brief discussion of atomic structure, the nuclear reaction, health effects of radiation, and nuclear reactors, can ‘cut to the chase’ by skipping to *The Case for Nuclear Power* (p. 21).

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## Background

### Atomic Structure

Many readers may already understand that all matter is composed of atoms and that atoms are composed of a nucleus itself comprising one or more of each of (positively charged) protons and (neutrally charged) neutrons, surrounded by a cloud of (negatively charged) electrons. A neat discussion of this is available as a three-and-a-half-minute Nova video (Kestin, undated). Elements in the Universe (carbon, oxygen, nitrogen, gold, nickel, etc., See Table 1) are different and exhibit different properties based on the number of protons present in the nucleus of the atom, the number of neutrons in the nucleus and the number of electrons surrounding that nucleus. An atom is neutral (in charge) if the number of negatively charged electrons outside the nucleus is equal to the number of positively charged protons inside. The *atomic number* of an atom is the number of protons while the *atomic mass* is the number of protons plus neutrons. Based on their atomic number, the 118 known elements are arrayed in ascending value in the Periodic Table of the elements (Table 1, modified here from Sharp and Bryner, 2022). The array is determined also by the arrangement of the electrons – a subject more complex to explain than the space available here allows.

1 H																	2 He
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba	* 	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86R Rn
87 Fr	88 Ra	** 	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og
Insert above	* 	57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Pu	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu	
Insert above	** 	89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr	

The elements depicted in Table 1 are:

**Row 1** H – Hydrogen; He – Helium

**Row 2** Li – Lithium; Be – Beryllium; B – Boron; C – Carbon; N – Nitrogen; O – Oxygen; F – Fluorine ; Ne – Neon

**Row 3** Na – Sodium ; Mg – Magnesium; Al – Aluminum; Si – Silicon; P – Phosphorus; S – Sulfur; Cl – Chlorine; Ar – Argon.

**Row 4** K – Potassium; Ca – Calcium; Sc – Scandium; Ti – Titanium; V – Vanadium; Cr – Chromium; Mn – Manganese; Fe – Iron; Co – Cobalt; Ni – Nickel; Cu – Copper; Zn – Zinc; Ga – Gallium; Ge – Germanium; As – Arsenic; Se – Selenium; Br – Bromine; Kr – Krypton.

**Row 5** Rb – Rubidium; Sr – Strontium; Y – Yttrium; Zr – Zirconium; Nb – Niobium; Mo – Molybdenum; Tc – Technetium; Ru – Ruthenium; Rh – Rhodium; Pd – Palladium; Ag – Silver; Cd – Cadmium; In – Indium; Sn – Tin; Sb – Antimony; Te – Tellurium; I – Iodine; Xe – Xenon.

**Row 6** Cs – Cesium; Ba – Barium; \* Hf – Hafnium; Ta – Tantalum; W – Tungsten; Re – Rhenium; Os – Osmium; Ir – Iridium; Pt – Platinum; Au – Gold; Hg – Mercury; Tl – Thallium; Pb – Lead; Bi – Bismuth; Po – Polonium; At – Astatine; Rn – Radon.

**Row 7** Fr – Francium; Ra – Radium; \*\*, Rf – Rutherfordium; Db – Dubnium; Sg – Seaborgium; Bh – Bohrium; Hs – Hassium; Mt – Meitnerium; Ds – Darmstadtium; Rg – Roentgenium; Cn – Copernicium; Nh – Nihonium; Fl – Flerovium; Mc – Moscovium; Lv – Livermorium; Ts – Tennessine; Og – Oganesson.

**Row 8** (\* - insert at Row 6 asterisk above) La – Lanthanum; Ce – Cerium; Pr – Praseodymium; Nd – Neodymium; Pm – Promethium; Sm – Samarium; Eu – Europium; Gd – Gadolinium; Tb – Terbium; Dy – Dysprosium; Ho – Holmium; Er – Erbium; Tm – Thulium; Yb – Ytterbium; Lu – Lutetium.

**Row 9** (\*\*insert at Row 7 asterisks above) Ac – Actinium; Th – Thorium; Pa = Protactinium; U – Uranium; Np – Neptunium; Pu – Plutonium; Am -Americium; Cm – Curium; Bk – Berkelium; Cf – Californium; Es – Einsteinium; Fm – Fermium; Md – Mendelevium; No – Nobelium; Lr – Lawrencium.

**Isotopes** of an element are atoms with the same number of protons (i.e., same atomic number) but a different number of neutrons, thus a different atomic mass.

The following discussion is modified largely from Fuge (2021). Most atomic isotopes are stable, meaning they do not emit nuclear particles. Some isotopes, however, are unstable, meaning they emit nuclear particles through a process known as ionizing radiation or radioactivity. An unstable atom continues to emit particles (or decay) until it reaches a stable configuration. The length of time taken to transform through loss of these particles into another species of atom (with a different number of protons and/or neutrons) is measured in terms of the half-life of the process. This refers to the length of time taken for half of the nuclei to decay to another isotope species (either the same or a different element depending on the number of protons retained). This half-life reflects both the rate of decay and the intensity of emitted radiation; these both halving. Note that the product of this decay may also be unstable and radioactive, and thus itself decay further, emitting more radiation in the process. Decay continues, and radiation is emitted, until a stable configuration is achieved.

An example from Britannica (undated) illustrates the process:

“The radioactive isotope cobalt-60 [27 protons; 33 neutrons], which is used for radiotherapy, has, for example, a half-life of 5.26 years. Thus, after that interval, a sample originally containing 8 grams of cobalt-60 would contain only 4 grams of cobalt-60 and would also emit only half as much radiation. After another interval of 5.26 years, the sample would contain

only 2 grams of cobalt-60. Neither the volume nor the mass of the original sample visibly decreases, however, the unstable cobalt-60 nuclei decay into stable nickel-60 nuclei [28 protons; 32 neutrons; during decay a neutron is changed to a proton], which remain with the still-undecayed cobalt.

Several kinds of radiation emissions can occur during the decay process:

**Alpha ( $\alpha$ )** radiation occurs when an atom loses two protons and two neutrons (interestingly, this alpha particle is equivalent to a Helium atom – see Table 1). Alpha particles cannot penetrate skin but are dangerous if ingested when they can cause cell damage.

**Beta ( $\beta$ )** radiation involves the emission of a high energy, high speed, negatively charged electron or a positron (a positively charged particle the size and mass of an electron). Because of its small mass, it can travel further in the air and can also penetrate a few centimeters into the skin and thus cause health problems - though these are more severe if the particles are ingested. Beta-decay normally occurs in nuclei that have too many neutrons to achieve stability.

**Gamma ( $\gamma$ )** radiation usually occurs with alpha or beta radiation but does not consist of particles. Rather, the radiation comprises photons of energy that have zero mass or charge and can therefore travel further. However, they can be stopped by high atomic mass materials such as lead or depleted uranium – hence the lead protective shields when we submit to radiation analysis.

**X Rays** are similar to gamma radiation but originate from the electron cloud. They are usually longer wavelength, with lower energy than gamma rays.

**Neutron radiation** occurs when a neutron is emitted from the nucleus, usually because of spontaneous or induced fission of the nucleus (see: the basic nuclear chain reaction under Nuclear Power Generation discussed below).

For additional discussion see Mirion (2015).

### ***Nuclear Power Generation***

The essence of the nuclear electricity generation process is a nuclear chain reaction (Figure 1) where the result of fission in one atom, induced by bombardment by a neutron, releases energy and neutrons that collide with other atoms and, in turn, cause them to undergo fission releasing further energy and so on. The basic atom involved in nuclear chain reaction process is a fissile (splitable) isotope of Uranium ( $U^{235}$ ). Ross (2022) offers a discussion of uranium and its history and role in nuclear reactors and nuclear weapons. When a free neutron collides with an atom of  $U^{235}$  (with a nucleus of 143 neutrons and 92 protons, hence the atomic mass of 235), it splits the nucleus and releases energy and neutrons. The most common fission products are atoms with mass numbers around 90 and 137 (NIH 2012) such as Strontium 90 and Cesium 137. The emitted neutrons then collide with other  $U^{235}$  atoms causing further nuclear fissions and so on...., hence the designation 'chain reaction.' The main products of the fission reaction (Figure 1) are listed in Table 2 accompanied by their half-lives and decay

products. The nuclear chain reaction emits energy as heat. In a nutshell, this heat is then used to boil water, generate steam, and subsequently drive a turbine that generates electricity.

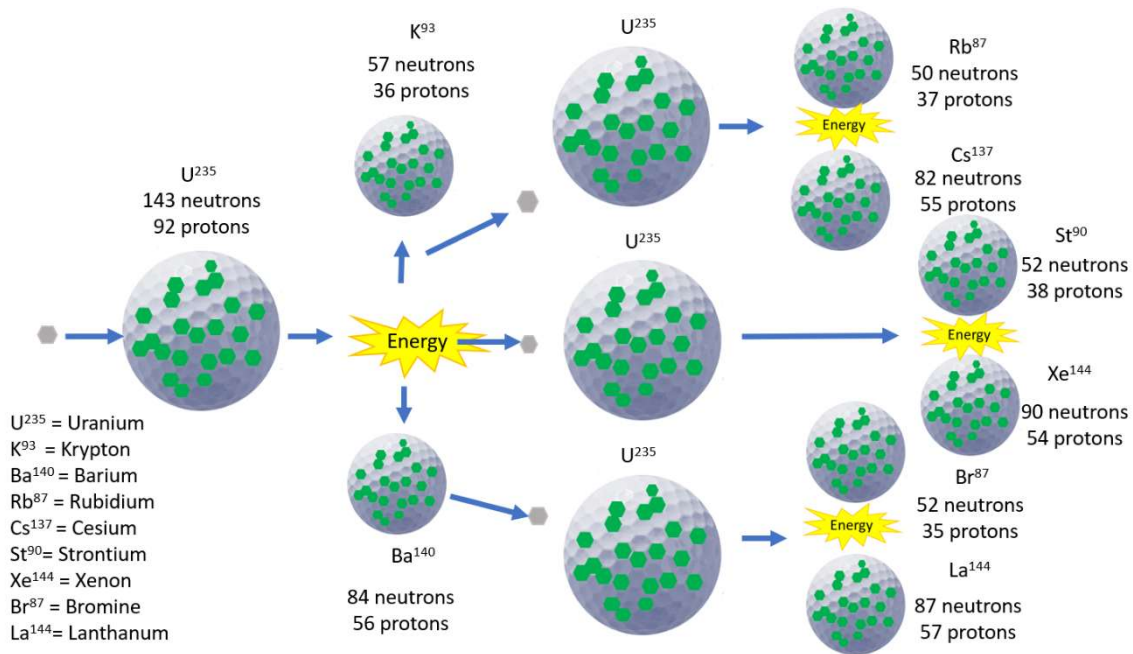


Figure 1. Example of the nuclear chain reaction. Adapted from Libretexts (2021)

Naturally occurring Uranium is rare, at 4 parts per million, and contains only about 0.7% the critical radioactive  $U^{235}$ , the rest being non-fissile  $U^{238}$  (meaning it cannot sustain a chain reaction). As a result, the fuel is enriched by gasification or by means of a centrifuge to achieve 3% to 5%  $U^{235}$ . After enrichment, and subsequent use in the reactor, the fuel becomes depleted Uranium since the density of  $U^{235}$  is reduced. This is far less radioactive than the initial fuel and only dangerous if inhaled or ingested. Uranium isotopes have variable half-lives:  $U^{238}$  - 4 billion years;  $U^{235}$  >700 million years;  $U^{234}$  about 25 thousand years. Depleted Uranium (Depleted  $UF_6$  undated) means that the substance emits very low-level radiation but does so for many years (see below). Additionally, this product is extremely tough physically and may be used as tank armor and as casing for armor-piercing shells.

Among the radioactive by-products of fission (Table 2), some are short-lived isotopes while others are very long-lived (Radioactivity, undated-a). Short-lived products vanish in under 5 years; medium-lived products last between 5 and 100 years (Radioactivity, undated-a), while (confusingly) long-lived products are defined as lasting longer than 30 years (Radioactivity, undated – b). This source also notes, however, that rapid decay rates, while leading to the conversion of the isotope to another species quickly, also emit more intense radiation than long-lived isotopes. While emitting for a much longer time, long half-life isotopes emit at a much lower intensity and are thereby less dangerous. A consequence of the rapid decay of the short and medium-lived isotopes is that after 500 years the radioactivity in spent fuel has decreased to a few hundredths or thousandths of the original hazard.

According to Sault (2020), the initial products of the fission of Uranium are: Krypton Kr92 and Barium Ba141; additional fission products are: Cerium Ce144, Praseodymium Pr144, Rubidium Rb87. The key members of the long-lived group (Table 2) are Strontium-90, Cesium-137, Plutonium-

<b>Table 2 The main radioactive products of nuclear fission and their half-lives. s = seconds, m = minutes, d = days, y = years, my = millions of years, by = billions of years.</b>				
Element – Isotope(s)	Half-life	Radiation	Decay Product	Notes
Iodine I <sup>131</sup>	8 d	Beta	Xenon Xe <sup>131</sup>	
Barium Ba <sup>140</sup>	12.8 d	- Beta	Lanthanum <sup>140</sup>	
Xenon Xe <sup>140</sup>	14 s	- Beta		Other Xenon isotopes have longer half-lives
Cerium Ce <sup>144</sup>	285 d	Beta	Praseodymium <sup>144</sup>	
Praseodymium Pr <sup>144</sup>	17.28 m	- Beta	Neodymium <sup>144</sup>	
Rubidium Rb <sup>87</sup>	48.8 by	Beta	Strontium <sup>87</sup>	
Rhodium Rh <sup>102</sup>	207d	Beta	Palladium <sup>106</sup>	
Promethium Pm <sup>147</sup>	2.6 y	Beta	Samarium <sup>147</sup>	
Strontium Sr <sup>90</sup>	30 y	Beta	Yttrium <sup>90</sup>	
Cesium Cs <sup>137</sup>	30 y	Beta	Barium <sup>137</sup>	
Plutonium <sup>239</sup>	24,360 y	Alpha	Uranium <sup>235</sup>	
Technetium Tc <sup>99</sup>	215,000 y	Beta	Ruthenium <sup>99</sup>	
Tin Sn <sup>126</sup>	230,000 y	Beta Gamma	Antimony <sup>126</sup>	
Selenium Se <sup>79</sup>	327,000 y	Beta	Bromine <sup>79</sup>	
Zirconium <sup>93</sup>	1.5 my	(low energy) Beta	Niobium <sup>93</sup>	Half-life 14y via low energy gamma rays to Niobium <sup>93</sup>
Cesium <sup>135</sup>	2.3 my	Beta	Barium <sup>135</sup>	
Palladium <sup>107</sup>	6.5 my	Beta	Rhodium <sup>106</sup>	Which decays to Silver
Iodine <sup>129</sup>	16 my	Beta Gamma	Xenon <sup>129</sup>	

239, Technetium-99, Tin-126, Selenium-79, Zirconium-93, Cesium-135, Palladium-107, and Iodine-129. This table also displays the radiation form and products. The result of their presence is that the radioactivity, though low intensity, reaches (decreases to) a plateau which will last hundreds of thousands of years.

<b>Table 3 The isotopes of Plutonium, their half-lives and decay products.</b>					
Isotope	Pu-238	Pu-239	Pu-240	Pu-241	Pu-242
Percentage	2%	53%	25%	15%	5%
Half-life (years)	88	24,000	6,560	14.4	374,000
Product	U-234	U-235	U-236	AM-241	U-238

Recall that the enriched Uranium fuel comprises 3 – 5% U<sup>235</sup>; the rest is U<sup>238</sup>. When the neutrons produced from fission of the former collide with the U<sup>238</sup> this atom is converted to Plutonium (Table 3) which exists as several

isotopes (with percentage, half-lives, and decay products as indicated).

Table 4 Uranium 238 decay chain and emissions (OpenLearn undated)

Radionuclide	Half-life
uranium -238	4.5 x 10 <sup>9</sup> years
α↓	
thorium -234	24.1 days
β↓	
protactinium -234	1.17 days
β↓	
uranium -234	2.45 x 10 <sup>5</sup> years
α↓	
thorium -230	7.5 x 10 <sup>4</sup> years
α↓	
radium -226	1600 years
α↓	
radon -222	3.82 days
α↓	
polonium -218	3.05 minutes
α↓	
lead -214	26.8 minutes
β↓	
bismuth -214	19.7 minutes
β↓	
polonium -214	1.6 x 10 <sup>-4</sup> seconds
α↓	
lead -210	22 years
β↓	
bismuth -210	5 days
β↓	
polonium -210	138 days
α↓	
lead -206	stable

The decay of radioactive elements can produce a chain of decay products, each with its own half-life to the next isotope in the chain. For example, the decay chain for Uranium 238, with emissions, is depicted in Table 4. Throughout the sequence, until the final stable isotope is achieved, potentially hazardous radioactive emissions occur.

During normal operations, nuclear power generators release small amounts of low-level radioactive isotopes (EPA 2022a). Additionally, Fast Company (2022) reports that: “the majority of reactors have leaked tritium, a radioactive form of hydrogen that can contaminate drinking water and, at high enough concentrations, cause cancer and genetic defects.”

### ***Controlling the Nuclear Reaction and Meltdown***

The chain reaction is regulated by control rods of boron, cadmium, hafnium, or other elements that absorb the neutrons and thus slow the rate of their collisions with uranium or plutonium. If the control rods are removed, the rate of neutron collisions increases, the fission rate increases and energy production and release increase (Libretexts 2021). In addition to the

control rods regulating the reaction, the fission reaction is cooled by water pumped around the reactor core (Matson 2011). Optimally, the heated coolant water is cooled and recycled back through the system. Sub-optimally, it is discharged into surrounding waters increasing their temperature and compromising aquatic wildlife species. Matson (2011) continues by explaining that without a steady supply of coolant, the reactor core may continuously emit heat and boil off the coolant. Ultimately, the fuel rods will no longer be immersed, thus allowing the fuel to heat and pool at the base of the reactor container. This hot puddle of fuel may then melt through the steel containment vessel and any additional barriers (hence ‘meltdown’) escaping into the outside world to expose the environment to their radioactive isotope decay products.

“After a severe accident such as a core meltdown (a kind of accident U.S. nuclear regulators call “beyond design basis”), a reactor may emit radiation into the environment, impacting all life and land around it...” (NRDC 2022).

## ***Health Effects of Radioactive Isotopes***

### ***General Effects.***

As with most toxins, the key issues with radiation are the strength of the exposure dose and the nature of the exposure (CDC 2021). Radiation can cause damage to the chromosomal DNA in the nucleus (the genetic material controlling cell development and function) either (a) by breaking DNA bonds and thus causing mutations (DNA reorganizations) that may be cancerous (thus, exposure is especially problematic as cells are dividing), or (b) by breaking bonds in the water surrounding the DNA and producing free radicals – unstable oxygen molecules that can damage tissue. Following radiation damage, cells may (i) repair the DNA and resume normal function, (ii) become permanently altered when there is a risk of cancer (uncontrolled cell division leading to tumors and leukemias, or (iii) be killed resulting in organ failure. Short term intense (high dose) radiation is more dangerous than long-term low-level radiation.

In a discussion of the risks of exposure to radionuclides, EPA (2021a) outlines the main variables imposing health risks:

- a) The energy emitted by the radiation,
- b) The type of radiation (alpha, beta, gamma x-rays discussed above),
- c) How often radiation is emitted,
- d) Whether exposure is external (source is outside the body) or internal (source has been ingested or inhaled) with the latter more serious,
- e) If ingested / inhaled, the rate at which the body metabolizes and eliminates the element,
- f) Whether the isotope accumulates in the body and how long it remains.

Because cell division is the most susceptible phase in the cell cycle, the same source (EPA 2021a) identifies children and fetuses as particularly vulnerable. This is largely because their cells are dividing rapidly.

While there are those who argue that no level of radiation is safe, the fact that we are constantly exposed to low levels of naturally occurring radiation both from terrestrial and cosmic sources leads some authorities (International Atomic Energy Agency – IAEA 1998-2022) to argue that: “...any exposure above the natural background radiation should be kept as low as reasonably achievable, but below the individual dose limits. The individual dose limit for radiation workers averaged over 5 years is 100 mSv), and for members of the general public, is 1 mSv per year.” Sievert (Sv) is the Standard International unit for measuring ionizing radiation dose, measuring the amount of energy absorbed in a human's body per unit mass, where mSv - milli Sievert or 1,000<sup>th</sup> of a Sievert. This is argued by the International Commission on Radiation Protection quoted by IAEA (1998-2022). Meanwhile, after indicating some years ago that there is no safe level of radiation (NAS 2005) more recently the National Academy of Sciences (NAS 2022) urged a \$100 million study of the effects of low-level radiation exposure. In the earlier study (NAS 2006) the conclusion was that women and children are more susceptible to ionizing radiation than healthy males. Unfortunately, it is the



latter who often stand as the reference case. This means that the sensitivity of the most vulnerable individuals is not the reference case.

Desbiolles *et al.* 2017 reported on the incidence of cancer in the vicinity of French nuclear power plants. They concluded that generally there was no trend except bladder cancer and that was attributed to two sites with confounding chemical pollution.

### ***Nuclear Isotope Hazards:***

As the U.S. Energy Information Agency acknowledges (EIA 2021) “A major environmental concern related to nuclear power is the creation of radioactive wastes such as uranium mill tailings, spent (used) reactor fuel, and other radioactive wastes. These materials can remain radioactive and dangerous to human health for thousands of years.”

“[The] residual radioactivity of fission products is due to a handful of long-lived isotopes. Key members of this group are technetium-99, cesium-135, iodine-129, palladium-107 and zirconium-93. Zirconium and palladium are chemically not very mobile, which leaves cesium, iodine and technetium as the most serious potential troublemakers.” (Radioactivity, undated b).

***Iodine 129***, with a half-life of 15.7 million years (EPA 2022b) is among 37 isotopes of iodine, only one of which (Iodine 127) is not radioactive. Most of this in our environment comes from nuclear testing in the 1950s.

***Iodine-131***, with a half-life of 8 days, is the most carcinogenic of the iodine isotopes and can cause burns to the eyes and skin. If ingested, it accumulates in the thyroid increasing the risk of cancer or other thyroid disorders (CDC 2018). Cattle eating grass contaminated with I-131 can pass it on to consumers in their milk.

***Cesium 135*** has a half-life of 2.3 million years while the half-lives of the other 37 isotopes of Cesium, besides the stable form (Cs<sup>133</sup>), range from a few days to a fraction of a second. Cs 134, with a half-life of two years, and Cs 137, with a half-life of 30 years are the hazardous isotopes (Chemeurope 1997-2022a,b) which in compounds can bioconcentrate in the body (Science Direct 2022).

***Technetium-99*** occurs naturally in the Earth’s crusts in very small quantities and besides being produced as a byproduct of nuclear reactions and thus present in nuclear waste, also occurs as a byproduct of nuclear weapons explosions; its half-life is 211,00 years. If ingested, this form of Technetium concentrates in the thyroid and gastrointestinal tract increasing the risk of cancer. Meanwhile, Technetium 99m is a short-lived variant used in medicine with a half-life of 6 hours (EPA 2021b).

***Palladium 107*** with a half-life of 6.5 million years is the second longest-lived of the 7 long-lived nuclear fission radiation products but is the least radioactive (Chemeurope 1997-2022c). It undergoes pure beta decay to silver.

***Zirconium 93*** has a half-life of 1.53 million years decaying via low energy beta radiation to radioactive Niobium 93, which in turn has a half-life of 14 years and decays to ordinary Niobium 93 while emitting gamma radiation (Chemeurope 1997-2022d).

***Plutonium 239***, formed when an U<sup>238</sup> atom captures a neutron from the fission process, has a half-life of 24,000 years. Approximately 1.15% of the spent fuel from nuclear generation is

Plutonium while over 1/3<sup>rd</sup> of the energy produced in a nuclear reactor is derived from Plutonium fission. Weapons grade Plutonium is a by-product of normal reactor operation after three years. Meanwhile, weapons grade Plutonium can be recovered from uranium fuel after 2 -3 months of use in a reactor (World Nuclear Association 2021). This suggests that promoting nuclear energy generation is, indeed, a likely contributor at least to the potential for nuclear weapons proliferation since it provides the fuel. The health risks from Plutonium result from its ingestion, entry via a wound, or inhalation, with the last exposure offering the greatest threat.

According to Jacoby (2020) uranium oxide pellets spend about 5 years in a nuclear reactor by which time their effectiveness declines and they are replaced with fresh fuel. The spent fuel is about 95% uranium, is thermally hot, and also contains a mixture of radioactive Plutonium ( $\text{Pu}^{244}$ ,  $\text{Pu}^{242}$ ,  $\text{Pu}^{239}$  with half-lives respectively of 82 million, 380, and 24 thousand years) and other fission products and actinides (a series of 15 radioactive metallic elements with atomic numbers from 89 – 103).

**Selenium 79** is not one of the 7 long-lived fission products but is problematic because it is the only one of the nine radioactive selenium isotopes resulting from nuclear fission to be of concern. This concern is based on its half-life of 65,000 years during which it emits beta particles that can be hazardous if ingested (hpschapters 2001).

### ***Types of Nuclear Reactors:***

Essentially, the operating principle of the nuclear power plant is no different than that of coal- or gas-powered electrical generation facilities (NRDC 2022) where water is heated to produce steam that drives turbines that rotate to generate electricity.

Lyman (2021) pointed out that the basic nuclear technology employed across the globe is the Light Water Reactor (LWR) which uses conventional water as the basic agent for cooling the core (the container where the nuclear chain reaction occurs). However, about 10% of reactors, he noted, replace light water with heavy water (in which the Hydrogen is replaced by Deuterium having the same atomic number as H, i.e., a single proton, but with a neutron added, thus doubling the atomic mass). Lyman (2021) also indicated that there have been efforts to promote so-called Advanced Reactors that are Non-Light Water Reactors (NLWRs). These replace the water with substances such as liquid sodium, helium gas, or even molten salts. Boiling Water and Pressurized Water reactors (discussed briefly below) are the prevailing types of LWR.

### ***Boiling-water Reactors (Figure 2)***

“In a boiling-water reactor, the reactor core heats water, which turns directly into steam in the reactor vessel. The steam is used to power a turbine generator” (EIA 2022a). This, in turn, generates electricity.

**Pressurized-water Reactors (Figure 3)** This reactor is described as follows (EIA 2022a): “In these generators the chain reaction described above in the reactor core heats water and keeps it under pressure to prevent the water from turning into steam. This hot radioactive water flows through tubes in a steam generator.” This source continues: “A steam generator is a giant cylinder filled with nonradioactive water (or clean water). Inside the giant water-

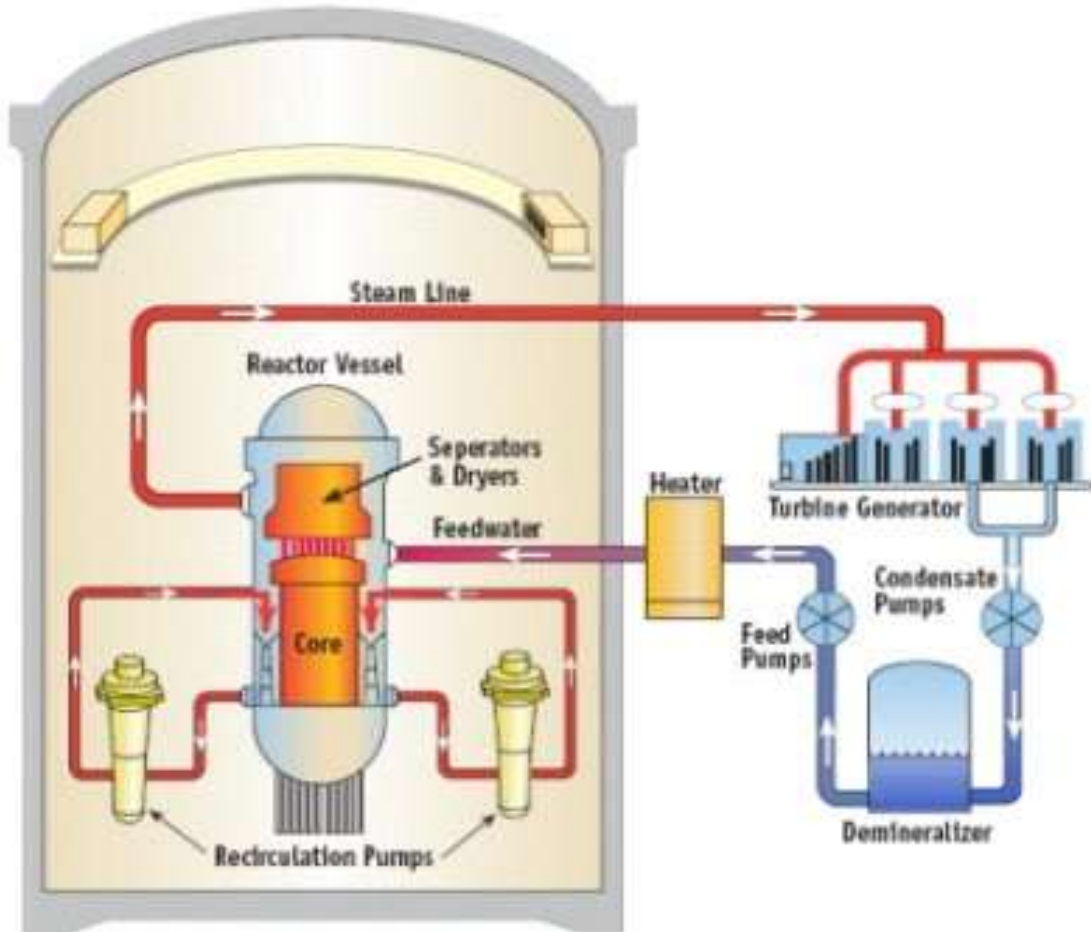


Figure 2. Diagram of a boiling-water nuclear reactor. Source: U.S. Nuclear Regulatory Commission (public domain) (EIA 2022a)

filled cylinder are thousands of tubes filled with the hot radioactive water from the reactor core that eventually bring the adjacent clean water to a boil and turn it into steam.” This comprises the nuclear heat exchange system. The radioactive water flows back to the reactor core to be reheated, and once reheated, returns to the steam generator” (EIA 2022a):

As of July 1, 2022, there were 92 nuclear reactors operating in the U.S. of which 61 were pressurized-water reactors (EIA 2022a). The NRDC (2022) provided a map displaying the locations of these reactors (Figure 4).

The basic nuclear power plant license issued by the Nuclear Regulatory Commission (NRC) is for 40 years, though operators can apply for an extension for another 20 years with no limit to the number of such re-applications (NRDC 2022). This NRDC commentary also points out that the average age of currently operating reactors is 40 years, while some plants are applying for an 80-year license and the NRC is discussing the possibility of 100-year permits.

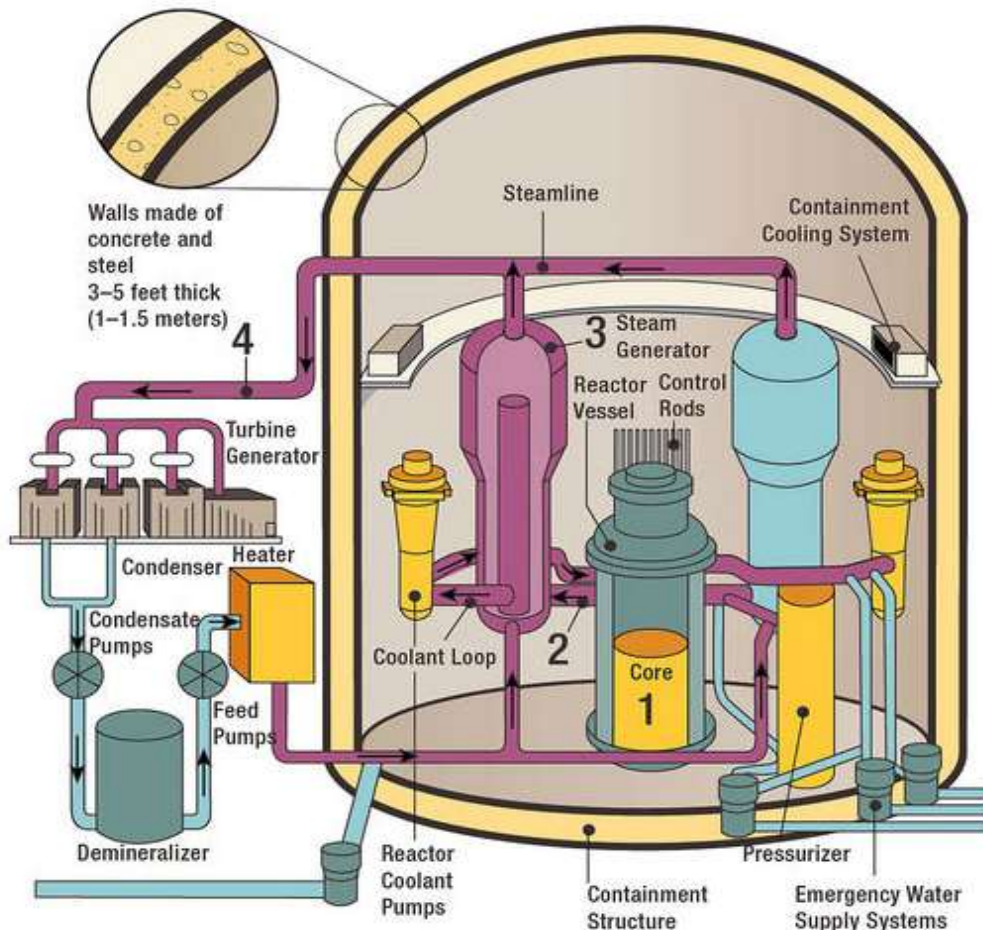


Figure 3 Diagram of a pressurized-water nuclear reactor. Source: U.S. Nuclear Regulatory Commission (public domain) (USNRC 2015)

*Legend*

- 1- The nuclear core inside the reactor vessel creates heat.
- 2- Pressurized water in the primary coolant loop carries the heat to the steam generator.
- 3- Inside the steam generator, heat from the primary coolant loop vaporizes the water in a secondary loop, producing steam.
- 4- The steam line directs the steam to the main turbine, causing it to turn the turbine generator, which produces electricity (USNRC 2015).

**Small modular reactors (SMRs)** According to the U.S. Energy Information Agency (EIA 2022a), these are about a third of the size of currently operational nuclear power plants while those currently under construction are simple and compact in design and can be assembled in a

factory and transported to the site. It is suggested that these features may allow more rapid nuclear power plant construction. Whereas large conventional nuclear reactors produce 700

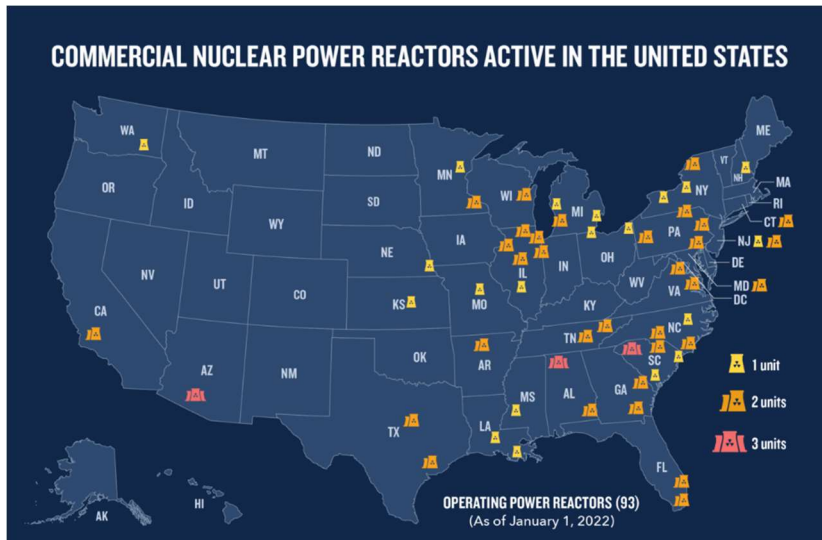


Figure 4. The location of nuclear reactors in the United States indicating multiple reactors at some sites. (NRDC 2022)

or more Megawatts (MW) of energy, SMRs operate at 300 MW, while the even smaller version Microreactors operate at up to 10 MW (Liou 2021). The same source suggests that SMRs require less fuel and also more infrequent refueling than conventional reactors. However, as of 2021, the International Atomic Energy Agency (Liou 2021) states “their economic competitiveness is still to be proven in practice ...”

According to the IAEA’s Liou (2021), SMRs are small reactors with a capacity to generate up to 300 MW per unit (about a third the capacity of traditional nuclear power plants) and produce low carbon electricity. The modular aspect refers to the fact that they are prefabricated off-site and then transported as a unit to the site. Because of their small footprint (area requirement), they can be sited where large conventional nuclear reactors are not possible. Because of their reduced cost in construction, they can be installed incrementally as energy (electricity) demand increases.

Liou (2021) also pointed out that: “The IAEA expects to publish a Safety Report on the applicability of IAEA safety standards to SMR technologies in 2022.” According to Donavan and Vives (2022) as of April 2022, the IAEA had completed its review of safety standards that would be applicable internationally and was expected to publish a report later in 2022. Liou (2022) then reported that IAEA’s Nuclear Harmonization and Standardization Initiative, charged with developing such standards, with a focus on SMRs first met in June. Since SMR development is occurring in many nations, the discussions involved 133 participants from 33 nations. It is not clear when the necessary codes and standards will appear.

Nakhle C. (2022) indicates that not only does it take on average eight years to build a nuclear power plant, more importantly the time between the decision and the commissioning can vary between 10 to 19 years. On the positive side, Nakhle (2022) argues, SMRs are not suitable for producing weapons-grade materials since uranium enrichment tends to be limited to 20 percent or less, so it is easier for them to comply with nonproliferation regulations. Furthermore, SMRs have reduced fuel requirements. She reports that the International Atomic Energy Agency (IAEA) stated that power plants based on SMRs may require refueling

every three to seven years, compared to every one to two years for conventional plants. Some SMRs are designed to operate for up to 30 years without refueling. The positive implication is that less frequent refueling decreases the risks inherent in transporting radioactive matter.

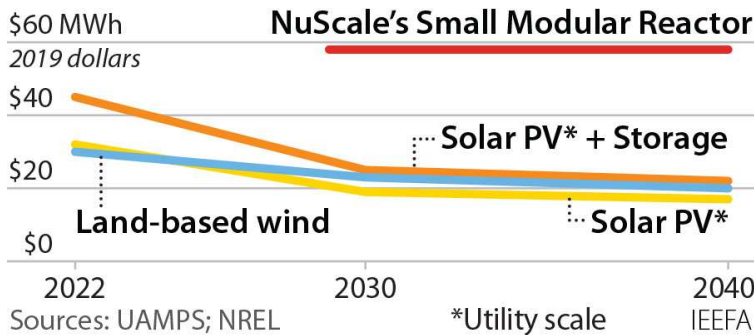
However, Nakhle (2022) acknowledged that at this stage, SMRs remain mostly a concept and that their economic competitiveness and general viability remain to be tested. She suggests that it is currently difficult to find reliable data demonstrating the commercial potential of this technology. As a result, it's difficult to offer a forecast.

An important measure of the merit of a technology is the levelized cost of energy (LCOE) of various sources of electricity. This metric incorporates the lifetime, capital cost, operations and maintenance expenses, fuel expenditures and energy production of a technology. Nakhle (2022) reports that one study found that SMRs are in fact the costliest option.

Farmer (2022) pointed out that despite their potential for use where conventional nuclear reactors are not possible, most current plans are to construct very small capacity SMRs on site alongside current reactors. Farmer (2022) also noted that Oregon's Nuscale SMR company

### NuScale's Power Price Target vs Renewables

The estimated price of power from its proposed SMR is much higher than the projected cost of alternatives



Sources: UAMPS; NREL \*Utility scale IEEFA

Figure 5. Cost estimates for Small Modular Reactor energy versus renewable energy. (IEEFA 2022)

plans to develop SMRs in Poland and build a plant in Idaho starting operations in 2029. However, Nuscale has attracted critique from the Institute for Energy Economics and Financial Analysis think tank. The institute (IEEFA 2022) reported that the Nuscale design was “too late and too expensive, too risky and too uncertain” compared to modern solar and wind renewables. Nuscale claims its SMR have a generating cost of \$58/MWh, through to 2040 with a

construction cost of \$3,000/kW. Meanwhile, the IEEFA (2022) cost analysis (Figure 5) reports that the comparable cost for solar and wind energy will be much lower by then.

Additionally, in reviewing the waste production from SMRs, Krall *et al.* (2022) concluded that in comparison to existing PWRs (Pressurized Water Reactors), SMRs will increase the volume and complexity of LILW (Low and Intermediate Level Waste) and Spent Nuclear Fuel. This increase of volume and chemical complexity will be an additional burden on waste storage, packaging, and geologic disposal.

The discrepancy regarding nuclear proliferation risk may be explained as Virgili (2020) indicates by the fact that some SMRs operate on < 20% enriched uranium, while some use > 20% enriched uranium.



They conclude that of three distinct SMR designs they assessed, that relative to a gigawatt-scale PWR reactor, these reactors “will increase the energy-equivalent volumes of Spent Nuclear Fuel, long-lived LILW, and short-lived LILW by factors of up to 5.5, 30, and 35, respectively.”

Small Modular Reactors seem to offer some advantages over the conventional nuclear reactor, but many questions remain. The evidence presented above suggests that there is little likelihood that SMRs will provide a satisfactory climate remedy.

***Advanced Non-Light Water Reactors (NLWRs)*** In reviewing Sodium-cooled fast reactors (SFRs), High temperature gas-cooled Reactors (HTGRs), and Molten salt-fueled reactors (MSRs) for the Union of Concerned Scientists, Lyman (2021) reported that the answer to the key question of whether these represent an advance over the conventional Light Water Reactors is: “Little evidence supports claims that NLWRs will be significantly safer than today's LWRs. While some NLWR designs offer some safety advantages, all have novel characteristics that could render them less safe.”

### ***The Thorium Option***

Several years ago (whatisnuclear 2007-2022a), the notion of a thorium-based fission process appeared on the scene as a safer and more reliable approach to nuclear generation than the conventional uranium process. However, it turns out that a number of myths were generated during that promotion (whatisnuclear 2007-2022b, Krahn and Worrall 2016) and the thorium option has since lost support.

According to the World Nuclear Association (WNA 2020), Thorium is a slightly radioactive element existing as Thorium-232. It is not usable directly in nuclear reactors but is fertile meaning not itself fissile but convertible into a fissile atom (NRC 2021). Thus, it must first be converted through irradiation to uranium-233, a process that parallels the conversion of Uranium 235 to Plutonium 239 in conventional reactors. The sources of this transformative irradiation are U-233, U-235 or Pu-239.

They (WNA 2020) identify the disadvantages of thorium as including: the fabrication cost to produce the fissile Plutonium; that U-233 is always contaminated with U-232 having a 69-year half-life and decaying to high gamma radiation emitters such as Thallium-208; that Thorium itself is difficult to recycle due to presence of Thallium-228 (an alpha emitter with 2-year half-life). However, some of these concerns, it is argued, can be overcome in a Liquid Fluoride Thorium Reactor.

Although the Thorium cycle is touted as less susceptible to promoting weapons proliferation, WNA (2020) notes that the U-233 is defined by the International Atomic Energy Agency in the same category as High Enriched Uranium (HEU) meaning (Zielinski 2012) it has been enriched to over 20% of the basic U-235. Naturally occurring Uranium-238, meanwhile, contains only about 0.7% U-235, an insufficient proportion to sustain a nuclear reaction much less create a bomb. Nuclear reactors need at least 3 – 4% U-235, while a bomb requires 90% U-235.

Meanwhile, among the advantages (WNA 2020) are that the Thorium nuclear cycle produces less waste than the conventional reactor.

Krzyzaniak & Brown (2019) argue that the claim that thorium reactors would be more economical than traditional uranium reactors because thorium is more abundant, has more energy potential and doesn't have to be enriched with the comment is false. They point out that the cost of uranium is a small fraction of the overall cost of nuclear energy since nuclear energy economics are controlled largely by the construction cost of the facility and thorium reactors are no cheaper.

Krzyzaniak & Brown (2019) also argue against the claim that thorium reactor waste is easier to deal with than current uranium reactor waste by pointing out that the thorium-uranium waste product has a similar radioactivity after 100 years and a greater radioactivity after 100,000 years. Meanwhile, in terms of the purported lower nuclear weapons proliferation risk, they report that by-products of the thorium cycle can be attractive for developing nuclear weapons.

In addition, NS Energy (2018) reports that while thorium is abundant, can be used in an array of reactor designs, offers reduced weapons proliferation opportunities compared to conventional nuclear generators, results in reduce hazardous waste production, and is safer to extract, it involves higher start-up costs, requires high temperatures to produce the thorium oxide than is required for producing uranium oxide, and can result in substantial emissions of gamma radiation from the U-232 present in irradiated thorium.

According to Pistilli (2022), since Thorium has been considered an excellent nuclear energy alternative for decades, it's hard to believe the safety and efficiency benefits have not led to more popular use of thorium reactors. She acknowledges, however, a major reason is that thorium-based reactors are still not economically viable for the most part while Uranium has benefited from decades of research, development and infrastructure thanks to its dual applications in weapons and energy during the Cold War. This research has allowed countries to establish protocols, infrastructure and knowledge bases that make uranium-based energy an easier option. She then concludes that at least for now, thorium reactors are unlikely to gain the upper hand over uranium oxide reactors. It's possible that thorium reactors could become more dominant in the future, but a lot of work will have to be done to get to that point.

In a discussion of the capacity for thorium to compete with uranium Dumé (2022) states: "Th-232 is of interest for nuclear power generation because it can easily absorb neutrons and transforms into Th-233. This new isotope emits an electron and an antineutrino within minutes to become protactinium-233 (Pa-233). This isotope, in turn, transforms into U-233, which is an excellent fissile material. Indeed, the fission of a U-233 nucleus releases about the same amount of energy (200 MeV) as that of U-235." However, she adds: "The U-233 produced at the end of the cycle is also difficult to handle, as it contains traces of U-232, which actively emits gamma radiation. While some researchers support the use of thorium as a fuel because its waste is more difficult to turn into atomic weapons than uranium, others argue that risks remain."



In terms of the capacity to generate weapons-grade material from the thorium reaction, Nuclear Matters (2020) argues that it is a possible pathway but “This process is rarely used because thorium (Th) as a nuclear fuel is less efficient than either natural uranium in a heavy-water reactor or low-enriched uranium in a light-water reactor.” It is also pointed out that “... the uranium-233 produced is less efficient as a fissile material than plutonium...”

The evidence suggests that the benefits of the Thorium cycle have been exaggerated and that this technology offers little advantage over the conventional uranium reaction. As a result of the negatives, the thorium option has more recently lost support.

### ***The cost and timeline for nuclear deployment***

Caroline Reiser, a staff attorney with Natural Resources Defense Council’s nuclear team, points out (quoted in NRDC 2022) that nuclear power plants “cannot compete economically with other low-carbon energy sources, like solar and wind, or with investments in energy efficiency.” Reiser continues: “while...advocates argue that nuclear power is important to decarbonizing the economy, it simply isn’t a solution to the climate crisis, especially in the time frame that we need to act.”

Jacobson (2019) underlined this point with the estimation: “New nuclear power plants cost 2.3 to 7.4 times those of onshore wind or utility solar PV per kWh, take 5 to 17 years longer between planning and operation, and produce 9 to 37 times the emissions per kWh as wind.” A 2022 report by the U.S. Energy Information Agency (EIA 2022b) identified the cost of nuclear electricity at some 3 – 4 times that of solar photovoltaic or wind-generated electricity. Interestingly, as an aside, ultra-supercritical coal-fired power plants with 90% carbon capture and storage (CCS – a technology promoted by coal companies as a way to maintain coal in the climate conscious energy mix. See, for example, GE undated) appear as expensive as nuclear power.

### ***Nuclear Power Plant Safety***

#### ***Nuclear Waste***

As the U.S. Energy Information Agency acknowledges (EIA 2021): “A major environmental concern related to nuclear power is the creation of radioactive wastes such as uranium mill tailings, spent (used) reactor fuel, and other radioactive wastes. These materials can remain radioactive and dangerous to human health for thousands of years.” This discussion continues: “The radioactivity of these wastes can range from a little higher than natural background levels, such as for uranium mill tailings, to the much higher radioactivity of used (spent) reactor fuel and parts of nuclear reactors.....” Nuclear waste is categorized as low-level waste and high-level waste.

Bemnet Alemayehu, a staff scientist with NRDC’s Climate & Clean Energy Program, quoted by NRDC (2022) stated: “Leading science holds to the linear no-threshold (LNT) model for radiation protection, which assumes that even very small doses of radiation can still increase the risk for cancer....”

By volume, most of the waste resulting from reactor operation is low-level waste (EIA 2021). Low-level waste comprises uranium mill tailings, and the tools, protective clothing, wiping cloths, and other disposable items that become contaminated with small amounts of radioactive dust or particles at nuclear fuel processing facilities and nuclear power plants...” while “High-level radioactive waste consists of irradiated, or spent, nuclear reactor fuel (fuel that is no longer useful for producing electricity)...” (EIA 2021).

In addition, once a nuclear plant has served its purpose for some decades, aged, and its useful life ended, it is closed down. This involves decommissioning during which the structure itself is dismantled comprising: “safely removing from service the reactor and all equipment that has become radioactive and reducing radioactivity to a level that permits other uses of the property (EIA 2021).

The Health Physics Society acknowledge that during normal operation, some (low-level) radioactivity is released (Radiation Answers 2022). However: “The radioactive material is held for a period of time to allow for the radioactivity level to decrease before being treated and/or released in a planned, monitored way.”

According to the Nuclear Information and Resource Service (NIRS undated): “A typical 1000-megawatt pressurized-water reactor (with a cooling tower) takes in 20,000 gallons of river, lake or ocean water per minute for cooling, circulates it through a 50-mile maze of pipes, returns 5,000 gallons per minute to the same body of water, and releases the remainder to the atmosphere as vapor. A 1000-megawatt reactor without a cooling tower takes in even more water--as much as one-half million gallons per minute. The discharge water is heated and contaminated with radioactive elements in amounts that are not precisely known or knowable but are biologically active.”

### ***Nuclear ‘Accidents’***

NRDC (2022) identifies the major historical nuclear incidents:

Since the emergence of nuclear power generation in the 1950s, the worst incident was at Chernobyl in 1986 (then part of the Soviet Union, now, infamously, in Ukraine). Explosions and fire destroyed one of the units following a power surge resulting in radiation reaching as far away as Sweden some 1,000 miles away. Amid a blanket of secrecy that made matters worse, the Soviet Government established an exclusion zone rather than cleaning up the mess. Thirty-one individuals died and 350,000 were evacuated.

Then, when Russian troops invaded Ukraine in 2022, as the 36<sup>th</sup> anniversary of the disaster approached, they captured Chernobyl on the first day, set up encampments, and exposed their troops to the radiation still present in that Exclusion Zone (Veranytsia and Veranytsia 2022).

This demonstrates that nuclear waste can never be considered secure. Kinley (2006) reported that: “Doses to the thyroid received in the first few months after the 1986 accident were particularly high in those who were children at the time and drank milk with high levels of

radioactive iodine. By 2002, more than 4,000 thyroid cancer cases had been diagnosed in this group, and it is most likely that a large fraction of these thyroid cancers is attributable to radioiodine intake.” Meanwhile, Gale (2021) reporting on longer-term consequences of the Chernobyl incident, stated: “First, there is no question there was an extraordinary increase in thyroid cancers in children and adolescents living in Ukraine, Belarus, and Russia, proximal to the accident.” And “There are about 7,000 excess thyroid cancers, an estimated 100-fold increased incidence, fortunately most not fatal.” Christodouleas *et al.* (2011) reported the array of radioactive isotopes released as a result of the Chernobyl accident as ranging from molybdenum 99 with a half-life of 67 hours to Plutonium 239 with a half-life of 24,400 years. These authors pointed out that there were “28 deaths related to radiation exposure in the year after the accident.”

There is considerable controversy about the number of deaths attributed to Chernobyl (Gray 2019) with an initial internationally recognized death toll of just 31 and UN estimates of only 50 deaths. However, in 2005, the U.N. predicted a further **4,000 might eventually die as a result of the radiation exposure (Blakemore 2019).**

The second worst global event followed a tsunami with 30-foot waves disabling the nuclear facility at the Fukushima Daiichi Nuclear Power Plant in Japan. Although the radiation released resulted in no immediate deaths, over 100,000 residents were evacuated, and the entire nation experienced radioactive fall-out. The economic cost of this disaster is likely to exceed \$200 billion. Christodouleas *et al.* (2011) noted that this incident resulted in the release of radioactive water into the ocean but argued that it diffuses rapidly with distance and decays over time.

Possibly the best-known incident in the United States remains the 1979 Three Mile Island partial meltdown which caused no immediate deaths but resulted in hundreds of thousands of individuals voluntarily evacuating. Clean-up took more than a decade and cost \$1 billion in 1993 dollars (over double that in today’s dollars). Evidence regarding short-term increase in cancer risk was inconclusive (Hatch *et al.* 1990). Of course, immediate deaths are not the only measure of health risk since the latency period between exposure and disease may be several years. Wing *et al.* (1997) reviewed earlier data and concluded that cancer risk had, indeed, increased downwind of the plant. Meanwhile, in a recent post (TMIA 2019) Wing was reported as stating: “Many earlier researchers, as well as government and industry officials, accept as fact that only small amounts of radiation were released into the atmosphere. But it is known that plant radiation monitors went off scale when the accident started. Plumes containing higher radiation could have passed undetected.” In addition, “I think our findings show there ought to be a more serious investigation of what happened after the Three Mile Island accident,” Wing said.

The nuclear power plants themselves are not the only hazard. In 1979, the same years as the Three Mile Island incident, the largest release of radioactivity in the U.S. occurred when a dam collapsed and released 1,000 tons of radioactive mine tailings into the Navajo Nation near Church Rock, New Mexico (Richards 2013) flowing past the homes of 1,700 Navajo residents. “...by the time uranium mining finally petered out in the early 1980s, hundreds of Indian

miners had died from lung diseases and cancers that physicians and secret U.S. Public Health studies linked to the miners' uranium exposure." This seems a perfect example of how frontline vulnerable Americans are ignored in debates over safety and health issues.

Events during the Russian invasion of Ukraine at the largest nuclear power plant in Europe (Zaporizhzhia) reveal serious dangers during military conflict (UN News 2022). Regardless of who is responsible for the shelling of the plant, we are seeing that nuclear power plants can be:

- designated as military targets,
- threatened with attack, or
- used as shields from which to launch offensive missile strikes.

The abuse of nuclear facilities by terrorists and in times of military conflict cannot be ignored, an especially critical problem if nuclear reactors are scattered around the planet in areas of the world exhibiting civil unrest and political instability. Promoting renewable energy generation facilities would be far safer than promoting nuclear generation in such locations.

Finally, the fact that conventional nuclear power plant operation can produce fuel for nuclear weapons, as discussed below, adds substantially to the risk of this technology.

### ***Nuclear Weapons Proliferation***

The product of conventional Heavy Water and Light Water nuclear reactors is Plutonium an atom that can be fabricated into the necessary fissile component for nuclear weapons. Schellenberger (2018) pointed out that "Of the 26 nations around the world that are building or are committed to build nuclear power plants, 23 have a weapon, had a weapon, or have shown interest in acquiring a weapon..." There is an obvious risk in promoting nuclear reactors as global sources of energy in that this inevitably places this technology and nuclear proliferation in the hands of potentially unstable nations across the globe, vastly increasing the risk of nuclear conflict

### ***The Price Anderson Act***

This Act, initially passed by Congress in 1957 and since renewed, provides the nuclear industry with a substantial subsidy that tilts the energy economic playing field in its favor.

As described by Holt (2018) the Price Anderson Nuclear Industries Indemnity Act authorizes the Nuclear Regulatory Commission to limit the liability of nuclear licensees from radiation damage to the public. This authority has been extended by Congress four times; it currently remains in effect until 2025. The Act requires nuclear generator owners (1) to carry insurance liability up to the current commercially available maximum, (\$450 million as of January 1, 2017), (2) for owners of 100-megawatt-and-above power reactors to contribute to an industry-wide fund which covers damages above \$450 million through a contribution by each nuclear reactor owner of up to \$121.3 million. As a result of the number of reactors liable for this payment, the total in this fund caps at \$12.4 billion but it is variable depending on the number of liable reactors. Damages above this amount would require Congressional action to

be funded, but there is no source for any such funds, so these would come from general revenue (i.e., the U.S. taxpayer).

It's worth being reminded that restoration of damage from the Fukushima incident is anticipated to cost over \$200 billion. Thus, if a Fukushima-like incident were to happen in the U.S. taxpayers would be responsible for over \$190 billion.

Holt (2018) also notes: "The Price-Anderson Act's limits on liability were crucial in establishing the commercial nuclear power industry in the 1950s. The nuclear power industry still considers them to be a prerequisite for any future U.S. reactor construction." We live in a society where individuals and businesses are, in many cases, required to carry insurance to cover damages should they be responsible for an accident. Surely, by the same token, an industry that claims to be safe should be required to negotiate insurance from private insurers. If that industry cannot persuade insurers to provide coverage, or afford the premiums levied, the message about safety of operations at nuclear reactors should be clear.

Polonsky & Eskelsen (2021) report that the Nuclear Regulatory Commission sought no changes in the liability plan when the Price Anderson Act was considered by Congress that year.

In 2021, the Department of Energy initiated a public comment period preparatory to a review of the act (Fork and Fowler 2021). These authors noted that the Price-Anderson Act is critical to nuclear suppliers' ability to manage their risk. In its report on recommendations regarding re-authorization, the Nuclear Regulatory Commission authors (USNRC 2021a) pointed out that the maximum coverage available per incident from this Act is \$13.4 billion. Since this total is based on the per reactor allocation to the fund, if the number of reactors decreases from the total (94 as of October 2020) the total funds available would decrease since each reactor contributes \$137.609 million to the fund. Despite the cost of the Fukushima incident estimation of over \$200 billion, this report relies only on the U.S. history of claims and the Three Mile Island example to conclude that the total available from the fund is sufficient to meet needs. This conclusion is offered despite the recognition that with nuclear plant retirements, this total number of reactors will likely drop, as will the amount in the fund. The key point not mentioned is that any accident cost beyond the fund that stands at a little over 1/10<sup>th</sup> the Fukushima cost will be covered by the U.S. taxpayer. In reviewing the NRC report, Lewis (2021) summarized it as follows: "The NRC ... does not recommend repealing or modifying any specific provisions, though it does make a few minor recommendations relating to the treatment of nonprofit operators and international coverage." The modifications include recommending a 10-year rather than a 20-year extension based on the anticipated deployment of advanced reactor designs.

It is not clear what Congressional action has occurred in relation to the proposed re-authorization.

### ***The Case for Nuclear Power***

Over the years of our awareness of the developing climate crisis, there have been repeated efforts to promote nuclear energy as a solution, or a major contributor to the solution. Some of these efforts have been clearly promoted by the nuclear industry itself apparently trying to

regain or increase its share of the energy market. Other efforts have been orchestrated by well-meaning individuals who genuinely see nuclear generation of electricity as having a substantial and beneficial role to play in fighting the climate crisis. Some have even interpreted the evidence to suggest it is essential if we are to succeed because we cannot achieve our global energy needs on renewable sources and storage alone.

Not surprisingly, the nuclear issue is not as clean as many proponents and opponents would have us believe.

Claims that we should promote nuclear energy seem to be based on several premises that deserve evaluation. The three main premises discussed here are as follows:

- 1) Clean renewable energy sources are inadequate to provide our energy needs.
- 2) Nuclear energy generation is free of greenhouse gas emissions.
- 3) Nuclear energy is safe.

To these, we might add three other considerations some of which have been discussed above:

- 4) There are new generation designs that are superior to the historic light water reactor design.
- 5) Failing to support aging nuclear reactors will likely result in their replacement by natural gas generators.
- 6) Economics and length of time to construction and function of nuclear facilities render them of questionable value in combatting the urgent climate crisis.

Evaluating these claims reveals a complex morass of sometimes conflicting evidence. This discussion will try to summarize the evidence as of December 2022.

### **1) *Clean renewable energy sources are inadequate to provide our energy needs***

In a recent report, Jacobson *et al.* (2022) analyzed the potential for genuine clean renewable energy – defined as energy: “that is both clean (emits zero health and climate affecting air pollutants when produced or consumed) and renewable (has a source that continuously replenishes the supply).” These energy sources mainly comprise Wind, Water and Solar energy with storage but include limited geothermal where available (see, for example, Solutions Project 2022). They do not include “fossil energy, bioenergy, non-hydrogen synthetic fuels, blue hydrogen, carbon capture, direct air capture, or nuclear energy, since each may result in a greater risk of air pollution, climate damage, and/or energy insecurity.” The only form of Hydrogen they considered was Green Hydrogen, that is Hydrogen produced by electrolysis using Wind, Water, or Solar (WWS) energy sources.

The authors noted that as of December 2021, 15 U.S. states, districts or territories had already established a 100% renewable energy goal (California, Connecticut, Hawaii, Maine, Nevada, New Jersey, New Mexico, New York, Oregon, Puerto Rico, Rhode Island, Virginia, Washington D.C., Washington State, and Wisconsin). If the nation were converted to 100% WWS energy by 2050/2051 some 4.7 million more long-term full-time jobs would be created than following a Business-as-Usual scenario involving accelerating fossil fuel use and consequent accelerating greenhouse gas emissions. The land area required for this system would be 0.84% of the

current national land surface, a figure which compares favorably with the current 1.3% of U.S. land surface utilized by the fossil fuel industry. Indeed, this represents but 65% of the current land area allocated to energy production.

A report by Bond *et al.* (2021) for Carbon Tracker concluded “With current technology and in a subset of available locations we can capture at least 6,700 PWh per annum (Picawatt (PWh) =  $10^{15}$  watts) from solar and wind, which is more than 100 times global energy demand. This opportunity presents itself because the cost of renewable energy has dropped. They also concluded that providing energy from solar alone would occupy only 0.3% of the land surface and that pricing will mean that: “fossil fuels will be pushed out of the electricity sector by the mid-2030s and out of total energy supply by 2050.”

Jacobson’s (2020) book, based on his Stanford University course, offers evidence that the world can be powered 100% on “clean, renewable wind-water-solar (WWS) energy and storage for everything.” He also argued that “The main obstacles appear to be social and political.”

In a study funded by the German Federal Environmental Foundation, Ram *et al.* (2019) concluded “A global transition to 100% renewable energy across all sectors – power, heat, transport and desalination before 2050 is feasible.” Furthermore, they indicated this is possible with “...e[E]xisting renewable energy potential and technologies, including storage.” They further conclude that a “sustainable energy system is more efficient and cost effective than the existing system, which is based primarily on fossil fuels and nuclear.” In assessing some proposed options, they state their proposed route “...achieves a cost decline without the reliance on high-risk technologies such as nuclear power and carbon capture and [storage] sequestration (CCS). A full energy transition to 100% renewable energy is not only feasible, but also cheaper than the current global energy system.” The obstacle, they suggest, is neither technological feasibility nor economic viability, but political will.

The evidence that wind, water and solar can provide our energy needs has been offered for several years (e.g., Jacobson *et al.* 2015). Although that paper received some criticism (Clack *et al.* 2017) the authors responded effectively (Jacobson 2017, Jacobson *et al.* 2017). Indeed, Jacobson (2017) also pointed out that “most of [the authors of the critique] have a history of advocacy, employment, research or consulting in nuclear power, fossil fuels or carbon capture. Through [The Solutions Project](#) (2022), Jacobson and his team have developed road maps for achieving 100% clean renewable energy economies in all 50 states and the nation as a whole as well as most other nations across the globe.

In a review of our energy future, energy and environment economist Mark Cooper (2021) argues that there is no benefit to continuing to subsidize and promote nuclear energy since the ‘nuclear renaissance’ has failed for multiple reasons. Amory Lovins (2021a) from RMI (formerly Rocky Mountain Institute) argued that proponents of nuclear power frame the issue incorrectly by focusing only on carbon and ignoring the issue of cost. Because clean renewables are cheaper, he argues, they displace less carbon per dollar than nuclear energy. Additionally, Lovins points out that most of the carbon emissions reductions in the U.S. to date have resulted from increased energy use efficiency and the increasing role of clean

renewable energy with nuclear playing a very minor role. Meanwhile, in a different discussion (Lovins 2021b), he addresses twelve myths that include some promoted by nuclear proponents. Notably, he points out, in response to the criticism that solar and wind electricity are intermittent, that the grid doesn't result in one generation source providing one user. Rather there is an array of generators covering a wide array of locations. Although it is certainly true that the sun sets and thus compromises solar generation overnight, as Torchinsky (2022) noted, a recent development in the technology shows promise of solar cells that can generate electricity at night by generating "electricity from the small difference in temperature between the ambient air and the solar cell itself." Meanwhile, a lull in wind turbine generation in one location can be compensated by generation elsewhere. Furthermore, the battery storage technology is advancing rapidly and compensating for intermittence involves more options than just batteries.

In addition, there are other techniques available for storing energy: already in use is the water storage system where water is pumped to a high elevation reservoir when energy is abundant, and then runs back down generating energy when the intermittent source is unavailable (e.g., EERE Undated a, EERE Undated b). An alternative, where local topography is not conducive to the water storage approach, energy can be used to raise a unit of mass (soil or rocks) which then can be lowered to emit the potential energy locked into the elevated mass (e.g., Moore 2021).

In an analysis of forecasts about the energy transition, Way *et al.* (2021) argued: "Most energy-economy models have historically underestimated deployment rates for renewable energy technologies and overestimated their costs." They concluded that: "compared to continuing with a fossil-fuel-based system, a rapid green energy transition will likely result in overall net savings of many trillions of dollars - even without accounting for climate damages or co-benefits of climate policy." Furthermore, they argue that because of the rapid decrease in renewable energy costs and their rapidly increasing deployment an energy future that relies on solar photovoltaics, wind, batteries and hydrogen electrolyzers is preferable because, "In contrast, a slower transition (which involves deployment growth trends that are lower than current rates) is more expensive and a nuclear driven transition is far more expensive." Again, we see evidence that clean renewable resources are expanding rapidly and offer a more cost-effective approach than promoting nuclear energy.

**Premise 1 Inference:** There seems abundant evidence, from many independent sources, that this premise for nuclear power, probably the most critical of all, is false. Rather, there is sufficient clean renewable energy to supply our needs. Since Premise 1 is the main premise underlying the argument that we need nuclear energy, its falsification constitutes a substantial blow to the entire argument that nuclear energy is necessary.

## **2) Nuclear energy generation is free of greenhouse gas emissions.**

When we undertake assessments of the climate impact of our activities, we must do more than examine day-to-day operations. We must examine the full life cycle emissions of that activity. In the case of solar and wind energy, this means examining the emissions that result from the extraction of materials and construction of the solar panels and wind turbines, plus



emissions resulting from their transport, installation and maintenance, and finally those resulting from their end-of-life removal and disposal. Fortunately, operation of these renewable generation sources is emissions-free. In the case of nuclear energy, this also means we must include the emissions resulting from the construction, operation, and decommissioning / disposal of the power plant, plus the emissions resulting from the extraction, processing and final disposal of the nuclear fuel. Fossil fuel-powered generation facilities are subject to the same assessment as the nuclear generator. Only when we are armed with these data are we able to make a legitimate comparison.

As an aside, it is worth noting that it's the failure to undertake full life cycle assessment that allows methane (natural gas) proponents to claim that fuel is 'the clean fossil fuel.' In fact, full life cycle assessment reveals that natural gas is as bad or worse than coal in terms of its full life cycle emissions. For more information on this, visit: [The Carbon Mistake](#), [The Natural Gas Conundrum](#), [What's Up With RNG?](#).

Over the years, several reports have been issued presenting life cycle emissions of various energy sources. These generally report emissions in grams (g) of carbon dioxide equivalent (CO<sub>2</sub>e) emitted per kilowatt hour (kwh) of electricity generated. Carbon dioxide equivalent is a measure of the warming impact of all gases assessed in terms of their equivalence to that of carbon dioxide – designated as 1. All analyses reveal that fossil fuels result in huge emissions in the many hundreds to thousands of g CO<sub>2</sub>e/ kwh. When Carbon Capture and Storage technology (CCS) is included, the emissions only drop slightly. This is not surprising, especially when one considers that CCS only addresses combustion emissions so upstream (extraction processing and distribution) emissions are untouched, as are those resulting from willful / unintended emissions from incomplete or inoperative flaring of methane. Given that substantial CO<sub>2</sub>e emissions from natural gas usage result from methane leakage upstream, or emissions resulting from unintended / willful incomplete or inoperative flaring, CCS can do little to reduce the climate pollution caused by this fuel. A National Energy Technology Laboratory report (Skone *et al.* 2015), for example, indicated that in the Appalachian Basin, 77% of the CO<sub>2</sub>e footprint of natural gas comprised methane with the majority of this resulting from distribution, transmission, and well completion. Presumably because the majority of CO<sub>2</sub>e emissions in the coal cycle result from combustion, imposing 90% carbon capture reduces coal-fired electricity emissions much more than it reduces gas-fired electricity generation emissions. Interestingly, this analysis reports the emissions from nuclear, hydroelectric, wind and solar in the range of 20 – 40 g CO<sub>2</sub>e/ kwh, while, at 250, geothermal is 6 to 10 times greater. Meanwhile, without CCS, coal is assessed at 1,205 g CO<sub>2</sub>e/ kwh, Petroleum at 1180 g CO<sub>2</sub>e/ kwh, and natural gas at 523 g CO<sub>2</sub>e/ kwh.

Jacobson (2020) also assessed the life cycle emissions of various fuels and concluded that nuclear power emits between 9 and 37 times more greenhouse gases (measured as CO<sub>2</sub>e) than wind power.

In an early literature review Sovacool (2008) summarized complete life cycle assessments of greenhouse gas emissions measured in g CO<sub>2</sub>e per kwh electricity generated. With numbers rounded, that report wind (onshore and offshore respectively) at 9 and 10, solar thermal

energy at 13 with photovoltaic solar at 32. Various nuclear reactor types averaged out at 66. At that time, the evidence suggests, nuclear was not assessed as equivalent to clean renewable energy sources in terms of emissions. These values compared with biomass (14 – 31) natural gas (443), diesel and oil (998) and coal (960, 1050) revealing how appalling all fossil fuels are by comparison. Additionally, since then, the warming impact of methane has been reevaluated time and again, and each time seems to be revealed as worse than previously thought. It's worth noting, also, that some of these assessments date from before the fugitive emissions (leakage) of methane in the extraction, processing, and transmission of natural gas were fully assessed and reported. These analyses have consistently demonstrated that natural gas is comparable to other fossil fuels in life cycle greenhouse gas emissions. Alvarez *et al.* (2018) reported that fugitive methane emissions produced a global warming impact equivalent to the combustion carbon dioxide emissions of the gas – negating the saving gas is often argued to exhibit because when we only consider combustion emissions, we find that emissions per unit of energy generated are lower for methane than for coal and oil. This would likely nearly double the natural gas impact reported above by Sovacool (2008). Indeed, Howarth (2015), a pioneer in the arena of life cycle assessment of the greenhouse gas emissions from natural gas, suggested that a main result of the inclusion of fugitive emissions is to reveal both shale-fracked and conventional natural gas produce a greater number of grams of carbon dioxide equivalent per mega Joule of energy generated than either coal or oil. Natural gas (methane) is not 'the clean fossil fuel.'

Chapter 7 of the Intergovernmental Panel on Climate Change evaluation of various energy sources (Bruckner *et al.* 2014) reported lifecycle assessments in a range of 675–1689 g CO<sub>2</sub>e/kWh electricity for coal. Corresponding ranges for oil and gas were 510–1170 g CO<sub>2</sub>e/kWh and 290–930 g CO<sub>2</sub>e/kWh<sup>14</sup>. They identified the ranges for lifecycle greenhouse gas emissions as 18–180 g CO<sub>2</sub>e/kWh for Photovoltaic panels, (Kim *et al.*, 2012; Hsu *et al.*, 2012), with 9–63 g CO<sub>2</sub>e/kWh for Concentrated Solar Power (Burkhardt *et al.*, 2012), and 4–110 gCO<sub>2</sub>e/kWh for nuclear power (Warner and Heath, 2012). Wind generation was graphed in the range of solar and nuclear, but the actual value was not reported.

Evans (2017) reported on CO<sub>2</sub>e emissions in a Carbon Briefs report from a publication by Pehl *et al.* (2017) using the same units (i.e., g CO<sub>2</sub>e per kwh electricity generated) as employed by Sovacool (2008) above and others, below. This assessment identified wind at 4, solar at 6 and nuclear at 4 g CO<sub>2</sub>e per kwh. Meanwhile, coal with Carbon Capture and Storage (109), natural gas with Carbon Capture and Storage (78), Hydro (97) and bioenergy (98) are all over an order of magnitude worse in terms of emissions.

The National Renewable Energy Laboratory (NREL 2021) reported life cycle assessments also measured in terms of g CO<sub>2</sub>e/kWh, though some were from much earlier studies dating from as far back as 2005 – presumably when no more recent study has been performed. Wind and nuclear tied at 13 g CO<sub>2</sub>e per kwh with concentrating solar power at 28 and photovoltaic panels at 43. Meanwhile, natural gas, oil, and coal respectively scored 486, 840, and 1001 g CO<sub>2</sub>e per kwh.

Jacobson (2019) further reported the 100 year life cycle assessment in terms of g CO<sub>2</sub>e/kWh as follows: rooftop solar 0.8 – 15.8, solar photovoltaic utility 7.85 – 26.9, concentrated solar power 8.43 – 25.2, onshore wind 4.8 – 8.6, offshore wind 6.8 – 14.8, hydroelectric 61-109, wave 26 – 38, tidal 14-36, nuclear 78 to 178 g, biomass 86 – 1,788, natural gas with CCUS at 230 – 481, and coal with CCUS 282-1,011 CCS where U represents Carbon Capture and Storage with Utilization. Unfortunately, the main utilization in CCUS of the gas is to promote further extraction of fossil fuels and thus generates further greenhouse gas emissions, which rather defeats the purpose.

Jacobson (2020) departed from the pattern of reporting nuclear generation as similar to solar and wind in terms of emissions per unit of energy generated. Rather, that author identified nuclear generation as producing between 9 and 37 times more CO<sub>2</sub>e and pollution than wind generation. The above data reveal that the range for solar means that nuclear could compare even less favorably.

While substantial differences exist among the studies, presumably based on slightly different methodologies, comparisons within studies reveal that nuclear generation, while consistently much lower than fossil fuels, is never a zero emissions process. It is not entirely clear if all life cycle assessments of the nuclear technology include decommissioning and waste storage, though they should. A study by Koltun *et al.* (2018) of a so-called fourth generation reactor (gas turbine technology with modular helium reactor GT-MHR) specifically included both decommissioning and waste treatment and revealed g CO<sub>2</sub>e/kWh of 15, well in line with the data reported above suggesting maybe these components are included.

**Premise 2 Inference:** While there seems little doubt that nuclear generation is a substantial improvement over coal, oil, and natural gas, at best, it appears to be right in line with the genuinely clean renewable sources of solar and wind. At worst, it simply may not achieve their low emissions so is no improvement over clean renewable sources. Furthermore, investment in nuclear energy would compete with investment in genuinely clean energy sources; every dollar spent on promoting nuclear energy is effectively a dollar subtracted from promoting renewable energy / storage. Meanwhile, as Matthews 2022 point out, the cost per megawatt hour of electricity generated (in 2021 dollars) is as follows: Solar \$36.49, Geothermal \$29.82, Onshore wind \$40.23 while hydro is \$64.27, ultra-supercritical coal is \$82.61, advanced nuclear is \$88.24, and biomass is \$90.17. The cost of nuclear power alone renders it non-competitive.

### **3) Nuclear energy is safe**

There exist two basic concerns regarding health and safety: one deals with the day-to-day operations (including waste production), the other with unpredictable events (whether human-induced or natural).

#### ***The Health and Safety Concern***

That nuclear fission poses potential health and safety risks is well known. As discussed above (Health Effects of Radioactive Isotopes), the problem with nuclear radiation from unstable isotopes is its proclivity for inducing cancer in exposed organisms. Exposure to the high energy radiation disrupts DNA in the nuclear chromosomes of the cells of exposed individuals, often causing cancerous hard or soft tumors.

As also discussed above, the risk posed by these isotopes depends largely on their half-lives: isotopes with short half-lives tend to emit intense radiation, while those with longer half-lives emit less intense radiation, but obviously do so over a much longer period.

It seems that the normal activity of a nuclear power plant will generally pose little threat to the environment though the heated water discharged from a plant may well disrupt local aquatic species. Problems arise, however, when normal activity is undermined – whether by human error, natural catastrophe (such as earthquakes and tsunamis), or terrorist/military assault.

However, it is worth remembering, as stated by NRDC (2022): “Current radiation protection standards are based on the premise that any exposure to radiation carries some risk, and that that risk increases directly with dose of exposure.” In a significant analysis of the literature, the National Academy of Sciences (NAS 2006) offered: “Epidemiologic studies ... show that exposure to low... radiation can lead to the age- and time dependent development of a wide range of tumor types that, in general, are not distinguishable from those arising in non-irradiated populations”

Richard Clapp, a retired professor from Boston University’s School of Public Health offered in a guest editorial in *Environmental Health Perspectives* (Clapp 2005) “Given the availability of alternative carbon-free and low-carbon options and the potential to develop more efficient renewable technologies, it seems evident that public health would be better served in the long term by these alternatives than by increasing the number of nuclear power plants in the United States and the rest of the world.”

Responding to an article promoting nuclear power in the energy mix, Larsen (2020) wrote: “...the proponents of nuclear power ...are overlooking the significant risks inherent in the technology and the fact that scaling up nuclear power would take too long and is too costly to be an effective climate solution.”

During the Russian invasion of Ukraine, invading forces took over the Ukrainian Chernobyl nuclear disaster site and apparently unwittingly exposed themselves to radioactive hazards. The invading Russian forces then targeted the largest nuclear power plant in Europe, in Zaporizhzhia, with shells and missiles before commandeering it. These events should be enough to alert everyone that nuclear facilities are sitting ducks for ignorant military or terrorist behavior and thereby pose an ongoing threat to citizens within many miles of the facility.

In responding to the claim of operational safety, frankly, it seems that no argument is really necessary except the single word: 'Zaporizhzhia.' It has long been suggested that a major threat posed by nuclear power plants is their exposure to terrorist action. Now, with the Russian invasion of Ukraine we have the perfect example of that threat as Putin's forces attacked the largest nuclear power plant in Europe and with its bombardment risked an outcome potentially equal to Fukushima. Given the number of unsettled regions around the globe, where civil unrest is possibly simmering just below the surface, the expansion of nuclear power, with its capacity to provide fuel to allow nuclear weapons proliferation, seems unwise at best. And if we acknowledge that this energy source is unnecessary, the notion of promoting nuclear power seems downright foolhardy.

It should be acknowledged, however, that maybe the Zaporizhzhia incident, emerging as a result of the Russian invasion of Ukraine is as strong an argument against a large, centralized power generation grid system as it is an argument against the energy resource used in that generation facility. The fact that the Ukrainian power plant is a nuclear facility simply compounds the risk.

### ***Extending Aging Reactor Licenses***

The number of nuclear reactors in the U.S. peaked in 1990 at 112 but has since declined as reactors age and competition from cheaper fossil fuels encroaches on profitability (Clemmer *et al.* 2018). They also point out that most U.S. reactors have 60-year operating licenses.

Lyman (2019) pointed out that as nuclear reactors age, "they require more intensive monitoring and preventive maintenance to operate safely." Lyman continues with the following comment: "Given that older reactors require more attention from the regulator, not less, it is perplexing that the NRC wants to scale back its inspections of the aging reactor fleet and its responses to safety violations." From a greenhouse gas emissions perspective, the problem is clear, as indicated by Rivero (2022): a quarter of nuclear capacity in advanced economies is expected to shut down by 2025. The problem is that low fossil fuel prices – exacerbated by the glut of natural gas – make ongoing nuclear operation less cost competitive. Because of the greater greenhouse gas emissions that result from natural gas (methane) electricity generation, the threat that nuclear capacity will be replaced by natural gas generation facilities is a serious problem.

The threat of this transition is exacerbated by lack of nuclear profitability. As Clemmer *et al.* (2018) pointed out, a third of U.S. nuclear power plants are either unprofitable (16) or scheduled to close (5) a number that means most reactor owners hold unprofitable plants. This lack of profitability would presumably be exacerbated if nuclear entities were required to fund their own insurance completely. They report the conflict between studies suggesting on the one hand that renewable resources and energy efficiency can achieve substantial emissions reductions while current reactors continue operation through their 60 years of licensing and studies, on the other hand, suggesting nuclear power will make a meaningful contribution to reducing emissions while assuming reactors will continue through and beyond their 60 years and the cost of new power plants will decline. However, as they note, predictions regarding nuclear costs have been notoriously underestimated by the industry.

The claim that nuclear power would be “too cheap to meter” (USNRC 2021b) was the first and most notable such overly optimistic claim. This should alert us to be skeptical about any claims of cheap energy offered by the nuclear industry. In an industry that is of questionable profitability where maintaining safety involves considerable expense, there exists considerable risk that safety will be sacrificed on the altar of profitability.

In assessing the future energy needs in this Union of Concerned Scientist (UCS) report, Clemmer *at al.* (2018) point out that while nuclear power accounted for 53% of low-carbon electricity in 2017, renewable energy was the fastest growing energy source. They also point out the need to replace high-emission sources with low-emission sources. For this reason, the UCS argues that a series of steps should be taken – including, but not limited to – support for maintaining aging nuclear power plants:

- State and Federal policies should support all low-carbon technologies
- Renewable energy and efficiency standards should be adopted
- A robust greenhouse gas emissions pricing system should be imposed
- Financial support is conditioned on consumer protection, safety requirements, and investment in renewable energy efficiency
- A low-carbon emissions standard should be established
- Power plant owners requesting financial assistance should be financially transparent and open their books
- Financial support for non-profitable power plants should be limited

While the UCS endorsement of nuclear generation was greeted enthusiastically by nuclear advocates, in promoting it they often overlook the array of caveats that accompany that endorsement. Especially overlooked are the correlated proposals to limit support for non-profitable nuclear power.

While the problem of aging nuclear plants being replaced by fossil fuel plants is real, the burgeoning availability of renewable energy and efficient battery storage suggest that, if undertaken, support for nuclear power should be employed on a short-term basis only and should not apply to new nuclear plants.

### **A Note About Fusion**

Anyone paying attention to science and energy news in late 2022 will have heard excited commentary about the progress towards nuclear fusion, a process that produces energy by combining light atoms (such as Hydrogen) rather than through nuclear fission of heavy atoms (such as Uranium and Plutonium). The catch, as Stallard (2022) describes the process, is the energy required to drive the process. Thus, the fusion process requires extremely high temperature (of the order experienced on the sun) accompanied by extremely high pressure. On the positive side, however, the energy is produced with a very low radioactive product and no greenhouse gases (so long as the energy used to drive the fusion process is clean).

DOE (2022) explains the process as follows: an atom of Tritium (an isotope of Hydrogen containing two neutrons and 1 proton) is fused with Deuterium (another isotope of Hydrogen

containing 1 neutron and 1 proton). In contrast the dominant and basic Hydrogen atom contains 1 proton and 1 electron but no neutrons (Brittanica 2022); the other isotopes are less common. The DOE (2022) notes that when the Tritium and Deuterium combine, they produce Helium with two protons and two neutrons meaning one neutron disappears. One neutron is expelled but some mass is lost, and this unit of mass is converted by the process into energy – hence the energy production of nuclear fusion. The reader might recall the Einstein principle that mass and energy are interconvertible (e.g., Fernflores 2019). Different atoms could be used in fusion, but the deuterium – tritium fusion releases most energy and can be conducted at lower temperatures than other fusions.

As Morelle (2022) points out, the experimental success is that the Lawrence Livermore National Laboratory in California has, for the first time, produced more energy than was consumed. To be sure, it was only enough to boil a few kettles, but it was a first step. From a purely scientific perspective, this is notable! The real question, however, is: can this become commercially and globally available within the time needed to address the climate crisis? This means within the next three decades. Unfortunately, turning this experimental procedure to a commercial venture that is readily available will take time. For many years, Morelle (2022) notes, 50 – 60 years has been the answer to the question: “how long?”

Morelle (2022) also notes an additional caution: despite the hype, we should recognize that the excess energy produced did not account for the energy needed to make the lasers work. When this is included, the procedure is no longer positive. While this fusion success is, indeed, a breakthrough, we should again recall the infamous 1954 quote (Terzic 2018) from then Chairman of the Atomic Energy Commission Lewis Strauss that nuclear fission energy would provide energy that is “too cheap to meter.” Yet, nuclear generation still requires huge taxpayer subsidy and even then often does not break even economically. The likelihood seems remote that nuclear fusion can become broadly available on a global scale to put much of a dent in the cause of the climate crisis in the time necessary.

### **Concluding Remarks and Summary**

In a sense, we can already assess what role nuclear energy might play in promoting a low emissions economy since among developed nations there exists a wide range in the degree to which electricity generation is driven by nuclear sources. If nuclear generation were to contribute substantially to lowering a nation’s emissions, this should be reflected in the relationship between nuclear emphasis among nations and the greenhouse gas emissions of those nations. Sovacool (2021) reported on this relationship and concluded that, in fact, greater emissions are associated with those nations that have more nuclear generation than those with less, a result contrary to the expectation if nuclear generation were to reduce emissions. Meanwhile, Sovacool (2021) also points out, nations with a greater emphasis on renewable generation exhibit lower GHG emissions than those utilizing nuclear generation. Rather than promoting emissions reductions, a nuclear emphasis seems merely to compete with and replace renewables. Maybe promoting nuclear energy psychologically encourages a ‘business as usual’ attitude among users that results in excessive energy utilization and undermines the encouragement of energy use efficiency and conservation. Since we know that there exists no totally benign energy source, actions that promote false solutions and

create the impression continued massive energy use is now acceptable are more dangerous than they might initially appear.

Sovacool (2021) adds the caveat that the lead time necessary to bring nuclear power plants online compared to that for renewable generation, indicates that, unlike renewables, nuclear generation is unlikely to be capable of addressing the climate crisis in the time necessary. The same author also notes that the International Energy Agency estimates the cost for installing sufficient nuclear capacity to address the problem globally would be \$4 trillion.

This analysis invites an obvious question: if we can provide our energy needs with genuinely clean renewable energy, why invest in, and subsidize, nuclear energy?

The 2014 Intergovernmental Panel on Climate Change AR5 report (Bruckner *et al.* 2014) states: “Continued use and expansion of nuclear energy worldwide as a response to climate change mitigation require greater efforts to address the safety, economics, uranium utilization, waste management, and proliferation concerns of nuclear energy use.”

In a review of the pathways mapped out to achieve a warming of 1.5°C above pre-industrial level by 2050, the IPCC 2018 (Masson-Delmotte *et al.* 2019, Chapter 2) noted “Nuclear power increases in most 1.5°C pathways with no or limited overshoot by 2050, but in some pathways both the absolute capacity and share of power from nuclear generators decrease “ (p.131) and “Some 1.5°C pathways with no or limited overshoot no longer see a role for nuclear fission by the end of the century...” (p. 131). It is noteworthy, however, that in those pathways including nuclear, the contribution of clean renewables is greater than that of nuclear so nuclear is seen, at best, as a minor augmentation to renewable energy.

Porritt (2021) noted that the drive for net zero emissions by 2050 has stimulated the nuclear industry and its proponents to promote that technology as a critical element in achieving the target. His conclusion was summed up simply as: “The problems they face are the same ones that have dogged the industry for decades: ever-higher costs, seemingly inevitable delays, no solutions to the nuclear waste challenge, security and proliferation risks.” This author also makes the important point that “Every kilowatt hour of nuclear-generated power will be a much more expensive kilowatt hour than one delivered from renewables plus storage.” This will compromise low-income Americans more than others.

We seem not to have made much, if any, progress in the nuclear arena over the last 15 years since the Green America (2006) identification of ten reasons to avoid the nuclear option:

- 1- Nuclear waste – waste remains radioactive for tens to hundreds of thousands of years
- 2- Nuclear proliferation - promoting nuclear energy programs increase the likelihood of proliferation of nuclear weapons enhancing the risk of these falling into hands of unstable governments.
- 3- National Security – the danger of nuclear plants serving as targets for aggressor military action and terrorism



- 4- Accidents – cancer rates among populations living in proximity to Chernobyl and Fukushima, especially among children, rose significantly in the years after the accidents,
- 5- Cancer risk – reports of increased cancer risk especially children among populations living near nuclear plants.
- 6- Energy production – in order to provide global energy (nuclear now supplied about 11%), 14,500 plants are required – accompanied by a massive increase in energy-intensive uranium mining.
- 7- Not enough sites – Since nuclear plants require a location near water supplies, there are simply insufficient sites globally to locate 14,500 plants.
- 8- Cost – Unlike cheap renewable energy, nuclear energy is extremely expensive.
- 9- Competition with renewables – Investing in nuclear energy would undermine funding for renewable energy.
- 10- Energy dependence of developing nations – the cost of nuclear would be beyond the capacity of many nations – making them dependent on developed nations for their energy sources. It's better to promote the far cheaper renewable energy alternative.

Although some evidence suggests that nuclear power may have a valuable role to play, the overall conclusion that the evidence suggests is that nuclear is not necessary, is too dangerous, could not be deployed in the time necessary to solve the crisis, does not lead to GHG emissions reduction, and merely competes for limited investment resources with genuine solutions. The evidence that nations with more versus less nuclear investment produce more greenhouse gas emissions also argues against nuclear energy as a solution to the climate crisis. Given the evidence that we have sufficient Wind, Water, and Solar resources plus storage, to serve our energy demand, we conclude that there is no good reason to divert time and financial resources away from such renewables to bolster a flagging nuclear industry.

SOCAN concurs with Mark Jacobson's (2021) assessment and his 7 reasons why nuclear energy is not the answer to solve climate change: "New nuclear power costs about 5 times more than onshore wind power per kWh. Nuclear takes 5 to 17 years longer between planning and operation and produces on average 23 times the emissions per unit electricity generated. In addition, it creates risk and costs associated with weapons proliferation, meltdown, mining lung cancer, and waste risks. Clean, renewables avoid all such risks."

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